i. Hydrogen + Chlorine → Hydrogen chloride

ii. Barium chloride + Aluminium sulphate → Barium sulphate + Aluminium chloride

iii. Sodium + water → Sodium hydroxide + water

iv. Sodium hydroxide solution (in water) reacts with hydrochloric acid solution (in water) to produce sodium chloride solution and water.

v. Calcium hydroxide + Carbon dioxide → Calcium carbonate + Water

vi. Zinc + Silver nitrate → Zinc nitrate + Silver

vii. Aluminum + Copper chloride → Aluminum chloride +Copper

viii. Barium chloride + Potassium sulphate → Barium sulphate + potassium chloride

ix. Potassium bromide (s) + Barium iodide (aq) → Potassium iodide (aq) + Barium bromide(s)

x. Zinc carbonate (s) → Zinc oxide (s) + Carbon dioxide (g)

xi. Magnesium (s) + Hydrochloric acid (aq) → Magnesium chloride (aq) + Hydrogen (g)

xii. Iron + Steam → Iron Oxide + Hydrogen

xiii. Silver bromide → Silver metal Bromine

xiv. Hydrogen peroxide→ Water + Oxygen

xv. Iron + Copper sulphate → Iron (II) sulphate + Copper

xvi. Zinc + Copper sulphate → Zinc sulphate + Copper

xvii. Lead + Copper chloride→ Lead chloride+ Copper

xviii. Potassium iodide + Chlorine → Potassium chloride Iodine

xix. Silver nitrate+Sodium chloride → Silver chloride+ Sodium nitrate

xx. Copper (Brown) +Oxygen→ Copper (II) oxide (Black)

xxi. Calciumoxide + Water → Calcium hydroxide

xxii. Sodium carbonate + Calcium chloride → Calcium carbonate + Sodium chloride

xxiii. Aluminium+ Copper chloride → Aluminiumchloride +Copper

xxiv. Zinc carbonate(s) → Zinc oxide(s) + Carbon dioxide(g)

xxv. Magnesium(s) + Hydrochloric acid(aq → Magnesium chloride(aq) + Hydrogen(g)

xxvi. Ferrous sulphate → Ferric oxide + Sulphur ioxide+ Sulphur trioxide

xxvii. Silver chloride → Silver + Chlorine

xxviii. Aluminium oxide → Aluminium + Oxygen

xxix. Silver nitrate+ Copper → Copper nitrate+ Silver

xxx. Sodium sulphate+ Barium chloride → Barium sulphate + Sodium chloride